



CaCO₃ recrystallization in saline and alkaline soils



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ABSTRACT

Are desert ecosystems a stable sink for atmospheric CO₂? Although the uptake of atmospheric CO₂ by soil during the nighttime is detected in deserts and some semi-deserts, the underlying mechanisms remain elusive. In order to determine the factors affecting the CO₂ fluxes into soil and to reveal the relationship between CO₂ fluxes and carbonate formation and recrystallization in saline and alkaline soils, four soils with contrasting salinity from two sites, Aksu and Yingbazar, along the Tarim River were analysed. Soils were incubated in ¹⁴CO₂ labelled atmosphere for 90 days at three CO₂ concentrations: 0.04%, 0.4% and 4%. The ¹⁴C activity was measured in soil water and air, as well as in carbonates after 2, 14 and 90 days. The ¹⁴C incorporation in CaCO₃ increased with corresponding ¹⁴C decrease remaining in the CO₂. The highest ¹⁴C incorporation into CaCO₃ (54%) was observed in the Yingbazar saline soil. The carbonate recrystallization rates increased logarithmically ($r^2 > 0.97$) with the CO₂ concentration. The average carbonate recrystallization rate from three sampling times was highest in the Yingbazar saline soil under 4% CO₂ ($8.45 \times 10^{-6} \text{ day}^{-1}$) and lowest in the Aksu alkaline soil under 0.04% CO₂ ($0.03 \times 10^{-6} \text{ day}^{-1}$). The average carbonate recrystallization rate increased with electric conductivity (corresponding to soil salinity) but decreased with pH ($R^2 = 0.99$ for 4% CO₂). The alkaline soils incorporated less ¹⁴C into CaCO₃ than the saline soils. At the highest CO₂ concentration of 4%, the full recrystallization period of the remaining primary carbonates was 10- to 100-fold shorter than at the lower CO₂ concentrations. Therefore, besides soil chemical parameters (e.g. pH, CaCO₃ content), CO₂ concentration (respiration of microorganisms and roots) is an important factor for CaCO₃ recrystallization and formation of pedogenic carbonates in desert soils. It is the most important factor for CaCO₃ recrystallization and so, for the formation of pedogenic carbonates.

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1. Introduction

Soils play an important role as carbon (C) sources (Ge et al., 2008; Wu et al., 2009) for changes in land use or as C sinks (Prentice and Fung, 1990) because of human activities. Desert and semidesert ecosystems are considered to be CO₂ sinks (Stone, 2008) because of the wide distribution of alkaline and CaCO₃ containing soils, which can be turned into carbonates. In west China, the average annual net ecosystem CO₂ exchange (NEE) was $-25.0 \pm 12.7 \text{ g C m}^{-2} \text{ yr}^{-1}$ from 2002 to 2012 in saline desert soil (Ma et al., 2014). The CO₂ uptake in alkaline and saline soils is estimated to be $0.3\text{--}3.0 \mu\text{mol m}^{-2} \text{ s}^{-1}$ (Xie et al., 2009). The Mojave desert soil in southwestern USA took up 102 ± 67 and $110 \pm 70 \text{ g C m}^{-2}$ in 2005 and 2006, respectively (Wohlfahrt et al., 2008). The C uptake of desert soils is therefore significant but may vary depending on soil properties.

The main mechanisms for the CO₂ flux into soil measured during the night in deserts and oasis cotton fields in Xinjiang, in the northwest of China, remains elusive (Stone, 2008). The CO₂ uptake at night cannot be attributed to photosynthesis, emphasizing the existence of abiotic processes for high CO₂ absorption (Ma et al., 2013; Xie et al., 2009). Using the $\delta^{13}\text{C}$ abundance in CO₂ (Walvoord et al., 2005) showed that CO₂ release from desert soil was derived from autotrophic and heterotrophic respiration components in the root zone and from calcite dissolution in deep water table settings. Via ¹³CO₂ dissolution in air-free water (Ma et al., 2014), desert soil are a sink for bicarbonate in shallow groundwater. Pedogenic carbonates are described as earthworm biospheroliths, rhizoliths and calcified roots, hypocotings, nodules, clast coating, calcretes and laminar caps (Zamanian et al., 2016b). Pedogenic carbonate-rhizoliths can form near the root zone as a function of water content, increased CO₂ concentration and dissolved CaCO₃ movement (Cramer and Hawkins, 2009; Gocke et al., 2010a; Gocke et al., 2014; Gocke et al., 2011a). The neoformation of pedogenic carbonates was discovered in saline-alkaline soil for 13 years irrigation (water contain Ca²⁺ and Mg²⁺) and fertilization (Bugchio et al., 2016). The secondary CaCO₃ precipitated in the soil with changing soil

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